

**APPENDIX A:**

**CHEMICAL VOLATILIZATION DATABASE**

Equations to Solve Mass Transfer Parameters in Database

General Equations Applied to All Applicable Sources:

<p>Chemical Stripping Efficiency = <math>\eta = \left(1 - \frac{C_1}{C_{1,in}}\right) \times 100\%</math></p>	<p><b>Variables</b>  <math>C_1</math> = chemical concentration in liquid phase out of system mg/L  <math>C_{1,in}</math> = chemical concentration in liquid phase entering system mg/L</p>	<p><b>Units</b> mg/L mg/L</p>																																				
<p>Mass Closure Estimate = % mass recovered = <math>\frac{V_l C_{l,2} + V_g C_{g,2} + Q_g \int_1^{t_2} C_g dt}{V_l C_{l,1} + V_g C_{g,1}}</math></p>	<p><b>Variables</b>  <math>V_l</math> = liquid volume L  <math>C_{l,1}</math> = chemical concentration in liquid phase at time 1 mg/L  <math>C_{l,2}</math> = chemical concentration in liquid phase at time 2 mg/L  <math>V_g</math> = headspace volume L  <math>C_{g,1}</math> = chemical concentration in gas phase at time 1 mg/L  <math>C_{g,2}</math> = chemical concentration in gas phase at time 2 mg/L  <math>Q_g</math> = ventilation rate of system L/min  <math>t_1</math> = time 1 min  <math>t_2</math> = time 2 min</p>	<p><b>Units</b> L mg/L mg/L L mg/L mg/L L/min min min</p>																																				
<p><math>k_g/k_l</math> Matrix Method</p>																																						
<table border="1" style="width: 100%; text-align: center;"> <thead> <tr> <th style="text-align: left;"><math>K_L A_i</math></th> <th>Chemical 1</th> <th>Chemical 2</th> <th>Chemical 3</th> <th>Chemical 4</th> <th>Chemical n</th> </tr> </thead> <tbody> <tr> <td>Chemical 1</td> <td>1</td> <td></td> <td></td> <td></td> <td></td> </tr> <tr> <td>Chemical 2</td> <td></td> <td>1</td> <td></td> <td></td> <td></td> </tr> <tr> <td>Chemical 3</td> <td></td> <td></td> <td>1</td> <td></td> <td></td> </tr> <tr> <td>Chemical 4</td> <td></td> <td></td> <td></td> <td>1</td> <td></td> </tr> <tr> <td>Chemical n</td> <td></td> <td></td> <td></td> <td></td> <td>1</td> </tr> </tbody> </table>	$K_L A_i$	Chemical 1	Chemical 2	Chemical 3	Chemical 4	Chemical n	Chemical 1	1					Chemical 2		1				Chemical 3			1			Chemical 4				1		Chemical n					1	<p><b>Variables</b>  <math>K_L A_i</math> = overall mass transfer value for chemical "i" L/min  <math>K_L A_j</math> = overall mass transfer value for chemical "j" L/min  <math>\Psi_i</math> = <math>(D_l/D_g)^{n1}</math>  <math>D_l</math> = liquid-phase diffusion coefficient for chemical "i" cm<sup>2</sup>/sec  <math>D_g</math> = liquid-phase diffusion coefficient for chemical "j" cm<sup>2</sup>/sec  <math>n_1</math> = liquid-phase power constant  <math>\Psi_g</math> = <math>(D_g/D_g)^{n2}</math>  <math>D_g</math> = gas-phase diffusion coefficient for chemical "i" cm<sup>2</sup>/sec  <math>D_g</math> = gas-phase diffusion coefficient for chemical "j" cm<sup>2</sup>/sec  <math>n_1</math> = liquid-phase power constant  <math>H_{ci}</math> = Henry's law constant for chemical "i" m<sup>3</sup> liq/m<sup>3</sup> gas  <math>H_{cj}</math> = Henry's law constant for chemical "j" m<sup>3</sup> liq/m<sup>3</sup> gas  <math>k_{gl}/k_{li}</math> = ratio of liquid- and gas-phase mass transfer coefficients for chemical "i"</p>	<p><b>Units</b> L/min L/min cm<sup>2</sup>/sec cm<sup>2</sup>/sec cm<sup>2</sup>/sec cm<sup>2</sup>/sec m<sup>3</sup> liq/m<sup>3</sup> gas m<sup>3</sup> liq/m<sup>3</sup> gas</p>
$K_L A_i$	Chemical 1	Chemical 2	Chemical 3	Chemical 4	Chemical n																																	
Chemical 1	1																																					
Chemical 2		1																																				
Chemical 3			1																																			
Chemical 4				1																																		
Chemical n					1																																	
<p>Three n x n matrices were filled with the following values:</p>																																						
<p><b>Matrix 1:</b> Ratio of measured <math>K_L A_i/K_L A_j</math> values for all chemicals and single experimental condition</p>																																						
<p><b>Matrix 2:</b> Ratio of predicted <math>K_L A_i/K_L A_j</math> values for all chemicals using an assumed <math>k_g/k_l</math> value in the following equation:</p>																																						
$\frac{K_L A_i}{K_L A_j} = \Psi_i \Psi_g \left\{ \frac{H_{ci}}{H_{cj}} \left[ 1 + \frac{k_{gl}}{k_{lj}} H_{cj} \right] \right\} \left\{ \frac{k_{gl}}{\Psi_1 + \Psi_g H_{ci} \left[ \frac{k_{gl}}{k_{lj}} \right]} \right\}$																																						
<p><b>Matrix 3:</b> Normalized residual between values in corresponding cells of Matrix 1 and Matrix 2. Each column and row of this matrix was added to find the total residual to be minimized. The value of <math>k_g/k_l</math> used to predict <math>K_L A</math> values in Matrix 2 was used to minimize total residual. Minimum residual value corresponded to "best" <math>k_g/k_l</math> value.</p>																																						
<p>Use <math>k_g/k_l</math> value and <math>\frac{1}{K_L A} = \frac{1}{k_L A} + \frac{1}{k_g A \bullet H_c}</math> to solve for liquid-phase mass transfer coefficient (<math>k_L A</math>) and gas-phase mass transfer coefficient (<math>k_g A</math>)</p>																																						

Solution Methods for Values of  $K_LA$  Referenced in Database

Method #

1

**Applicable Sources:** Shower, Flow-through Bathtub

**Model Equation:** 
$$C_{l,out} = C_{l,in} \exp\left(-\frac{K_LA}{Q_l}\right) + \left(\frac{C_g}{H_c}\right) \left(1 - \exp\left(-\frac{K_LA}{Q_l}\right)\right)$$
  
 Predicts  $C_{l,out}$

**Variables**

		<b>Units</b>
$C_{l,out}$ =	chemical concentration in liquid-phase exiting system	mg/L
$C_{l,in}$ =	chemical concentration in liquid-phase entering system	mg/L
$K_LA$ =	overall mass transfer coefficient multiplied by interfacial area	L/min
$Q_l$ =	system liquid flowrate	L/min
$C_g$ =	chemical concentration in gas-phase of system at time t	mg/L
$H_c$ =	Henry' law coefficient of chemical	(L <sub>liq</sub> /L <sub>gas</sub> )

**Solution Technique:** Solve for  $K_LA$  by minimizing normalized residual between measured  $C_{l,out}$  and predicted  $C_{l,out}$ :

$$\left[ \frac{(C_{l,out,measured} - C_{l,out,predicted})}{C_{l,out,measured}} \right]^2$$

2

**Applicable Sources:** Shower, Flow-through Bathtub

**Model Equation:** 
$$C_g = \frac{a}{b} + \left(C_{g,in} - \frac{a}{b}\right) \exp(-bt)$$
  
 Predicts  $C_g$

where: 
$$a = \frac{\left(Q_l C_{l,in} \left(1 - \exp\left(-\frac{K_LA}{Q_l}\right)\right)\right) + Q_g C_{g,ii}}{V_g}$$

$$b = \frac{\left(\frac{Q_l}{H_c}\right) \left[\left(1 - \exp\left(-\frac{K_LA}{Q_l}\right)\right) + Q_g\right]}{V_g}$$

**Variables**

		<b>Units</b>
$C_g$ =	chemical concentration in gas-phase	mg/L
$C_{g,in}$ =	chemical concentration in gas-phase at start of experiment	mg/L
$Q_l$ =	system liquid flowrate	L/min
$C_{l,in}$ =	chemical concentration in liquid-phase entering system	mg/L
$K_LA$ =	overall mass transfer coefficient multiplied by interfacial area	L/min
$Q_g$ =	system ventilation rate	L/min
$V_g$ =	headspace volume	L
$H_c$ =	Henry's law constant of chemical	(L <sub>liq</sub> /L <sub>gas</sub> )

**Solution Technique:** Solve for  $K_LA$  by minimizing normalized residual between measured  $C_g$  and predicted  $C_g$ :

$$\left[ \frac{(C_{g,measured} - C_{g,predicted})}{C_{g,measured}} \right]^2$$

3

**Applicable Sources:** Washing Machine Fill Cycle, Filling Bathtub**Model Equations:**Predicts  $C_l$  and  $C_g$ 

Discretized Equations:

$$C_l^{n+1} = \left[ \frac{Q_l C_{l,in}}{V_l^n} - \frac{Q_l C_l^n}{V_l^n} - \frac{K_L A C_l^n}{V_l^n} + \frac{K_L A C_g^n}{V_l^n H_c} \right] \Delta t + C_l^n$$

$$C_g^{n+1} = \left[ \frac{-Q_g C_g^n}{(V_t - V_l^n)} + \frac{Q_l C_g^n}{(V_t - V_l^n)} + \frac{K_L A C_l^n}{(V_t - V_l^n)} - \frac{K_L A C_g^n}{(V_t - V_l^n) H_c} \right] \Delta t + C_g^n$$

2nd Order Runge-Kutta Technique:

$$C^{n+1} = C^n + \frac{\Delta t}{2} \left\{ f(t^n, C^n) + f[t^n + \Delta t, C^n + \Delta t f(t^n, C^n)] \right\}$$

**Solution Technique:** Solve for  $K_L A$  by minimizing normalized residual between measured  $C_l^{n+1}$  or  $C_g^{n+1}$  and predicted values at associated time steps.

Note: Liquid-phase concentration equation is dependent on gas-phase concentration

$$\left[ \frac{(C_{l,measured} - C_{l,predicted})}{C_{l,measured}} \right]^2$$

**Variables**

$C_l^n$	chemical concentration in liquid phase at time step n	mg/L
$C_l^{n+1}$	chemical concentration in liquid phase at time step n+1	mg/L
$V_l^n$	liquid volume at time step n	L
$V_t$	total volume of system	L
$K_L A$	overall mass transfer coefficient multiplied by interfacial area	L/min
$Q_l$	system liquid flowrate	L/min
$C_{l,in}$	chemical concentration in liquid-phase entering system	mg/L
$H_c$	Henry's law constant of chemical	(L <sub>liq</sub> /L <sub>gas</sub> )
$C_g^n$	chemical concentration in gas phase at time step n	mg/L
$C_g^{n+1}$	chemical concentration in gas phase at time step n-1	mg/L
$Q_g$	system ventilation rate	L/min
$\Delta t$	differential time step	min

4

**Applicable Sources:** Dishwasher, Washing Machine Wash/Rinse Cycle, Handwashing Dishes in Kitchen Sink, Bathing**Model Equation:**Predicts  $C_l$ 

$$C_l = C_{l,0} \left[ \exp\left(-\frac{D}{2}t\right) \cosh\left(\left(\sqrt{\frac{D^2}{4} - E}\right)t\right) + \left(\frac{BF}{Z} + \frac{EC_{l,0}}{Z} - \frac{DC_{l,0}}{2}\right) \left[ \frac{1}{\left(\sqrt{\frac{D^2}{4} - E}\right)} \exp\left(-\frac{D}{2}t\right) \sinh\left(\left(\sqrt{\frac{D^2}{4} - E}\right)t\right) \right] \right]$$

where:

$$Z = \frac{K_L A}{V_l} \quad B = \frac{K_L A}{V_l H_c} \quad X = \frac{K_L A}{V_g}$$

$$Y = \frac{Q_g}{V_g} + \frac{K_L A}{V_g H_c} \quad D = Z + Y \quad E = ZY - BX$$

$$F = ZC_{g,0} + XC_{l,0}$$

**Variables**

$C_l$	chemical concentration in liquid phase	mg/L
$C_{l,0}$	initial chemical concentration in liquid phase	mg/L
$K_L A$	overall mass transfer coefficient multiplied by interfacial area	L/min
$V_l$	system liquid volume	L
$H_c$	Henry's law constant of chemical	(L <sub>liq</sub> /L <sub>gas</sub> )
$V_g$	system headspace volume	L
$Q_g$	system ventilation rate	L/min
$C_{g,0}$	initial chemical concentration in gas phase	mg/L
$t$	time	min

**Solution Technique:** Solve for  $K_L A$  by minimizing residual between measured  $C_l$  and predicted  $C_l$ :

$$\left[ \frac{(C_{l,measured} - C_{l,predicted})}{C_{l,measured}} \right]^2$$

5 **Applicable Sources:** Dishwasher, Washing Machine Wash/Rinse Cycle, Handwashing Dishes in Kitchen Sink, Bathing

**Model Equation:**  $C_g = C_{g,0} \exp\left(-\frac{D}{2}t\right) \cosh\left(\left(\sqrt{\frac{D^2}{4} - E}\right)t\right) + \left(F - \frac{DC_{g,0}}{2}\right) \left[ \frac{1}{\left(\sqrt{\frac{D^2}{4} - E}\right)} \exp\left(-\frac{D}{2}t\right) \sinh\left(\left(\sqrt{\frac{D^2}{4} - E}\right)t\right) \right]$   
 Predicts  $C_g$

where:  $Z = \frac{K_L A}{V_l}$        $B = \frac{K_L A}{V_l H_c}$        $X = \frac{K_L A}{V_g}$

$Y = \frac{Q_g}{V_g} + \frac{K_L A}{V_g H_c}$        $D = Z + Y$        $E = ZY - BX$

$F = ZC_{g,0} + XC_{l,0}$

**Variables**

$C_g$ =	chemical concentration in gas phase	mg/L
$C_{g,0}$ =	initial chemical concentration in gas phase	mg/L
$K_L A$ =	overall mass transfer coefficient multiplied by interfacial area	L/min
$V_l$ =	system liquid volume	L
$H_c$ =	Henry's law constant of chemical	(L <sub>liq</sub> /L <sub>gas</sub> )
$V_g$ =	system headspace volume	L
$Q_g$ =	system ventilation rate	L/min
$C_{l,0}$ =	initial chemical concentration in liquid phase	mg/L
$t$ =	time	min

**Solution Technique:** Solve for  $K_L A$  by minimizing residual between measured  $C_g$  and predicted  $C_g$ :

$$\left[ \frac{(C_{g,\text{measured}} - C_{g,\text{predicted}})}{C_{g,\text{measured}}} \right]^2$$

6 **Applicable Sources:** Kitchen Sink (with recirculating batch reactor experimental design)

**Model Equation:**  $C_l = C_{l,0} \exp(-K_L a t)$

Predicts  $C_l$

**Variables**

$C_l$ =	chemical concentration in liquid phase at time $t$	mg/L
$C_{l,0}$ =	initial chemical concentration in liquid phase	mg/L
$K_L a$ =	overall mass transfer coefficient multiplied by interfacial area/liquid volume	min <sup>-1</sup>
$t$ =	time	min

**Solution Technique:** Slope of best curve fit of  $\ln(C_l/C_{l,0})$  vs. time equal to  $K_L a$

Chemical stripping efficiencies also solved using best fit equation to measured liquid data according to:

$$C_l = C_{l,0} \exp\left(-\frac{t}{\theta_H} \eta\right)$$

**SHOWER DATABASE**

**STUDY:** Howard and Corsi  
**Study year:** 1997  
**Solution Methods:** Method 1 for acetone, ethyl acetate, toluene, ethylbenzene, cyclohexane  
**Assumptions:** None  
**Comments:** Values of  $\eta$  and  $K_{LA}$  are averages of values determined for three separate time periods within experiment.  
 Mass closure based on liquid standard curve created using well-dissolved tracer bag.

\* = unable to be determined with available data, n/m = not measured

Entry #	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
1	Spray Type =	coarse	$H_c @ 21\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_{LA}$	$k_fA$	$k_gA$	$k_g/k_l$	Mass Closure	
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	21	Acetone	0.0010	$1.1 \times 10^{-5}$	0.11	38	34 - 37	0	0.036	6.3	1.8	13	1986	153	99
	Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0042	$9.5 \times 10^{-6}$	0.092	26	20 - 24	0	0.068	15	2.9	7.3	1111	153	100
	Gas Flowrate (L/min) =	370	Toluene	0.24	$9.1 \times 10^{-6}$	0.085	4.5	1.8	0	0.041	61	8.8	9	1380	153	90
	Shower Stall Volume (L) =	1745	Ethylbenzene	0.27	$8.4 \times 10^{-6}$	0.077	6.1	2.3	0	0.029	62	8.9	9.1	1395	153	77
	Person Present =	No	Cyclohexane	6.5	$9.0 \times 10^{-6}$	0.088	1.8	0.63	0	0.0069	65	9.6	9.6	1468	153	78
2	Spray Type =	fine	$H_c @ 22\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_{LA}$	$k_fA$	$k_gA$	$k_g/k_l$	Mass Closure	
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	22	Acetone	0.0011	$1.1 \times 10^{-5}$	0.11	42	35 - 40	0	0.041	8.4	3.0	16	3519	223	97
	Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0044	$9.5 \times 10^{-6}$	0.092	22	16 - 20	0	0.081	15	4.0	8.1	1807	223	104
	Gas Flowrate (L/min) =	343	Toluene	0.25	$9.1 \times 10^{-6}$	0.085	7.0	2.3	0	0.057	68	11	11	2434	223	82
	Shower Stall Volume (L) =	1745	Ethylbenzene	0.27	$8.4 \times 10^{-6}$	0.077	7.3	2.3	0	0.032	68	11	11	2384	223	66
	Person Present =	No	Cyclohexane	6.6	$9.0 \times 10^{-6}$	0.088	3.3	0.88	0	0.0097	73	12	12	2652	223	68
3	Spray Type =	coarse	$H_c @ 21\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_{LA}$	$k_fA$	$k_gA$	$k_g/k_l$	Mass Closure	
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	21	Acetone	0.0010	$1.1 \times 10^{-5}$	0.11	42	36 - 39	0	0.035	9.1	1.4	8.6	1723	200	98
	Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0044	$9.5 \times 10^{-6}$	0.092	26	19 - 22	0	0.057	20	2.3	5.1	1030	200	98
	Gas Flowrate (L/min) =	360	Toluene	0.24	$9.1 \times 10^{-6}$	0.085	7.0	2.6	0	0.029	63	6.2	6.4	1274	200	82
	Shower Stall Volume (L) =	1745	Ethylbenzene	0.26	$8.4 \times 10^{-6}$	0.077	7.5	2.8	0	0.011	63	6.0	6.2	1234	200	64
	Person Present =	No	Cyclohexane	6.3	$9.0 \times 10^{-6}$	0.088	3.0	1.0	0	0.0079	66	6.5	6.5	1305	200	77
4	Spray Type =	fine	$H_c @ 22\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_{LA}$	$k_fA$	$k_gA$	$k_g/k_l$	Mass Closure	
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	22	Acetone	0.0011	$1.1 \times 10^{-5}$	0.11	43	37 - 41	0	0.037	9.3	1.5	8.8	1720	195	98
	Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0044	$9.5 \times 10^{-6}$	0.092	31	22 - 26	0	0.083	20	2.5	5.3	1031	195	101
	Gas Flowrate (L/min) =	360	Toluene	0.25	$9.1 \times 10^{-6}$	0.085	6.1	2.2	0	0.036	64	6.4	6.5	1275	195	92
	Shower Stall Volume (L) =	1745	Ethylbenzene	0.28	$8.4 \times 10^{-6}$	0.077	5.5	2.0	0	0.018	63	6.2	6.3	1232	195	76
	Person Present =	No	Cyclohexane	6.7	$9.0 \times 10^{-6}$	0.088	2.0	0.67	0	0.0063	66	6.7	6.7	1309	195	85

5	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
	Spray Type =	coarse	$H_c @ 35\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	35	Acetone	0.0022	$1.1 \times 10^{-5}$	42	34 - 38	0	0.077	13	2.8	14	1548	111	97
	Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0077	$9.5 \times 10^{-6}$	24	16 - 19	0	0.13	27	5.5	12	1322	111	102
	Gas Flowrate (L/min) =	379	Toluene	0.37	$9.1 \times 10^{-6}$	6.4	2.0	0	0.056	68	11	11	1223	111	77
	Shower Stall Volume (L) =	1745	Ethylbenzene	0.54	$8.4 \times 10^{-6}$	8.0	2.5	0	0.032	68	11	11	1188	111	68
	Person Present =	No	Cyclohexane	10	$9.0 \times 10^{-6}$	2.6	0.63	0	0.0093	75	13	13	1439	111	66

  

6	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
	Spray Type =	fine	$H_c @ 34\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	34	Acetone	0.0021	$1.1 \times 10^{-5}$	40	34 - 38	0	0.071	11	3.4	16	2095	131	99
	Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0074	$9.5 \times 10^{-6}$	24	15 - 18	0	0.12	28	6.9	14	1852	131	100
	Gas Flowrate (L/min) =	354	Toluene	0.35	$9.1 \times 10^{-6}$	5.3	1.3	0	0.053	75	13	14	1776	131	94
	Shower Stall Volume (L) =	1745	Ethylbenzene	0.51	$8.4 \times 10^{-6}$	4.6	1.1	0	0.031	75	13	13	1708	131	81
	Person Present =	No	Cyclohexane	9.6	$9.0 \times 10^{-6}$	1.6	0.35	0	0.0054	77	14	14	1786	131	74

  

7	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
	Spray Type =	fine	$H_c @ 34\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	34	Acetone	0.0021	$1.1 \times 10^{-5}$	40	32 - 37	0	0.071	12	3.7	15	2316	153	99
	Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0074	$9.5 \times 10^{-6}$	25	15 - 19	0	0.13	29	6.7	13	1945	153	102
	Gas Flowrate (L/min) =	373	Toluene	0.36	$9.1 \times 10^{-6}$	5.2	1.3	0	0.058	74	12	13	1930	153	90
	Shower Stall Volume (L) =	1745	Ethylbenzene	0.52	$8.4 \times 10^{-6}$	3.8	1.0	0	0.031	74	12	12	1855	153	78
	Person Present =	No	Cyclohexane	9.8	$9.0 \times 10^{-6}$	0.71	0.17	0	0.0065	77	13	13	1950	153	85

  

8	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
	Spray Type =	coarse	$H_c @ 36\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	36	Acetone	0.0024	$1.1 \times 10^{-5}$	41	32 - 36	0	0.054	16	2.2	11	1169	110	96
	Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0080	$9.5 \times 10^{-6}$	24	14 - 17	0	0.081	32	3.8	8.2	901	110	99
	Gas Flowrate (L/min) =	364	Toluene	0.38	$9.1 \times 10^{-6}$	5.6	1.5	0	0.036	74	8.4	8.6	949	110	98
	Shower Stall Volume (L) =	1745	Ethylbenzene	0.57	$8.4 \times 10^{-6}$	5.3	1.4	0	0.023	73	8.2	8.3	917	110	86
	Person Present =	No	Cyclohexane	10	$9.0 \times 10^{-6}$	1.9	0.46	0	0.0054	76	8.6	8.6	943	110	77

  

9	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
	Spray Type =	fine	$H_c @ 35\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	35	Acetone	0.0022	$1.1 \times 10^{-5}$	42	33 - 37	0	0.079	14	2.3	9.6	1380	143	102
	Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0077	$9.5 \times 10^{-6}$	24	14 - 18	0	0.12	33	4.7	9.0	1292	143	108
	Gas Flowrate (L/min) =	371	Toluene	0.36	$9.1 \times 10^{-6}$	6.0	1.6	0	0.048	73	8.1	8.3	1189	143	88
	Shower Stall Volume (L) =	1745	Ethylbenzene	0.53	$8.4 \times 10^{-6}$	5.6	1.5	0	0.029	72	7.9	8.0	1139	143	92
	Person Present =	No	Cyclohexane	10	$9.0 \times 10^{-6}$	1.8	0.43	0	0.0060	75	8.4	8.4	1203	143	81



Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	fine	H <sub>c</sub> @ 34 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	8	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	34	Acetone	0.0021	1.1 x 10 <sup>-5</sup>	0.11	40	32 - 37	0	0.081	15	2.5	11	1507	138	103
Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0074	9.5 x 10 <sup>-6</sup>	0.092	24	14 - 16	0	0.13	36	5.3	10	1443	138	106
Gas Flowrate (L/min) =	367	Toluene	0.36	9.1 x 10 <sup>-6</sup>	0.085	5.4	1.3	0	0.047	77	9.2	9.3	1291	138	88
Shower Stall Volume (L) =	1745	Ethylbenzene	0.53	8.4 x 10 <sup>-6</sup>	0.077	4.5	1.1	0	0.027	75	8.8	8.9	1227	138	75
Person Present =	No	Cyclohexane	9.9	9.0 x 10 <sup>-6</sup>	0.088	0.84	0.17	0	0.0057	80	9.9	9.9	1366	138	80

**STUDY:** Keating, McKone, and Gillett

**Study year:** 1997

**Reference:** *Atmospheric Environment, Vol. 31, No. 2, 1997, pp. 123-130.*

**Solution Methods:** Method 1

**Assumptions:** C<sub>g,in</sub> = 0

**Comments:** C<sub>g,end</sub> values picked off graph presented in paper.

C<sub>l,out</sub> and C<sub>l,in</sub> were measured in triplicate at 2, 4, and 8 minutes at the drain and shower nozzle, respectively.

Resulting measurements at each sample location, respectively, were not found to be statistically different, so values listed are averages for three sample times.

Results for studies 11, 12, 14 and 16 are based on three different experiments.

Results for studies 13 and 15 are based on two different experiments. (Individual results not given in paper - only averages).

Liquid flowrate value of 3.5 L/min is average of range of values given in paper (3.1 L/min to 3.8 L/min).

\* = unable to be determined with available data, n/m = not measured

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	"spray"	H <sub>c</sub> @ 40 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	10	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	40	Chloroform	0.37	9.7 x 10 <sup>-6</sup>	0.10	64	11	0	0.31	83	6.7	*	*	*	52
Liquid Flowrate (L/min) =	3.5	Chloroform	0.37	9.7 x 10 <sup>-6</sup>	0.10	57	20	0	0.31	65	3.8	*	*	*	74
Gas Flowrate (L/min) =	195														
Shower Stall Volume (L) =	1530														
Person Present =	No														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	"jet"	H <sub>c</sub> @ 40 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	10	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	40	Chloroform	0.37	9.7 x 10 <sup>-6</sup>	0.10	56	18	0	0.44	68	4.2	*	*	*	88
Liquid Flowrate (L/min) =	3.5														
Gas Flowrate (L/min) =	195														
Shower Stall Volume (L) =	1530														
Person Present =	No														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	1800 μm	H <sub>c</sub> @ 35 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	10	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	35	Chloroform	0.29	9.7 x 10 <sup>-6</sup>	0.10	57	6.4	0	0.58	89	9.1	*	*	*	84
Liquid Flowrate (L/min) =	3.5														
Gas Flowrate (L/min) =	195														
Shower Stall Volume (L) =	1530														
Person Present =	No														

14		Operating Conditions	Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	1800 $\mu\text{m}$		$H_c @ 45\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure
Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
Liquid Temperature ( C) =	45	Chloroform	0.47	$9.7 \times 10^{-6}$	0.10	59	5.1	0	0.48	91	9.7	*	*	*	67
Liquid Flowrate (L/min) =	3.5														
Gas Flowrate (L/min) =	195														
Shower Stall Volume (L) =	1530														
Person Present =	No														

15		Operating Conditions	Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	300 $\mu\text{m}$		$H_c @ 35\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure
Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
Liquid Temperature ( C) =	35	Chloroform	0.29	$9.7 \times 10^{-6}$	0.10	60	14	0	0.53	77	5.5	*	*	*	87
Liquid Flowrate (L/min) =	3.5														
Gas Flowrate (L/min) =	195														
Shower Stall Volume (L) =	1530														
Person Present =	No														

16		Operating Conditions	Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	300 $\mu\text{m}$		$H_c @ 45\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure
Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
Liquid Temperature ( C) =	45	Chloroform	0.47	$9.7 \times 10^{-6}$	0.10	51	10	0	0.54	80	6.0	*	*	*	95
Liquid Flowrate (L/min) =	3.5														
Gas Flowrate (L/min) =	195														
Shower Stall Volume (L) =	1530														
Person Present =	No														

**STUDY:** Giardino and Andelman  
**Study year:** 1996  
**Reference:** *Journal of Exposure Analysis and Environmental Epidemiology*, Vol. 6, No. 4, 1996, pp. 413-423  
**Solution Method:** Method 1  
**Assumptions:**  $C_{g,in} = 0$ .  
 No  $C_{l,out}$  values were reported, assumed it equal to  $(1-\eta) \times C_{l,in}$ , where  $\eta$  is given in paper.  
 $C_{l,out}$  remained relatively constant for entire experiment.  
**Comments:**  $C_{l,in}$  is reported average value for ten-minute experiment. Thus,  $\eta$  is an average value for ten-minute period.  
 \* = unable to be determined with available data, n/m = not measured

17		Operating Conditions	Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	unknown		$H_c @ 27\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure
Duration (min) =	23	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
Liquid Temperature ( C) =	27	Trichloroethene	0.46	$9.4 \times 10^{-6}$	0.084	0.49	0.160	0	0.020	67	6.3	*	*	*	121
Liquid Flowrate (L/min) =	5.0														
Gas Flowrate (L/min) =	88														
Shower Volume (L) =	1500														
Person Present =	No														



Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	unknown	$H_c @ 33\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =	19	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	33	Trichloroethene	0.62	$9.4 \times 10^{-6}$	0.084	0.52	0.130	0	0.055	74	20	*	*	*	124
Liquid Flowrate (L/min) =	10														
Gas Flowrate (L/min) =	28														
Shower Volume (L) =	1500														
Person Present =	No														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	unknown	$H_c @ 32\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =	19	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	32	Trichloroethene	0.59	$9.4 \times 10^{-6}$	0.084	0.88	0.30	0	0.033	66	12	*	*	*	107
Liquid Flowrate (L/min) =	9.9														
Gas Flowrate (L/min) =	225														
Shower Volume (L) =	1500														
Person Present =	No														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	unknown	$H_c @ 25\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =	19	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	25	Trichloroethene	0.41	$9.4 \times 10^{-6}$	0.084	0.79	0.330	0	0.058	59	12	*	*	*	112
Liquid Flowrate (L/min) =	9.9														
Gas Flowrate (L/min) =	33														
Shower Volume (L) =	1500														
Person Present =	No														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	unknown	$H_c @ 26\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =	11	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	26	Chloroform	0.18	$9.7 \times 10^{-6}$	0.10	0.63	0.360	0	0.011	44	3.2	*	*	*	116
Liquid Flowrate (L/min) =	5.0														
Gas Flowrate (L/min) =	70														
Shower Volume (L) =	1500														
Person Present =	No														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	unknown	$H_c @ 42\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =	11	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	42	Chloroform	0.35	$9.7 \times 10^{-6}$	0.10	0.58	0.220	0	0.016	62	5.3	*	*	*	131
Liquid Flowrate (L/min) =	5.0														
Gas Flowrate (L/min) =	65														
Shower Volume (L) =	1500														
Person Present =	No														



Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	unknown	H <sub>c</sub> @ 30 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure
Duration (min) =	14	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)
Liquid Temperature ( C) =	30	1,2-Dibromo-	0.01	7.6 x 10 <sup>-6</sup>	0.056	1.6	1.3	0	n/m	17	1.2	*	*	*
Liquid Flowrate (L/min) =	5.0	3-chloropropane												
Gas Flowrate (L/min) =	308													
Shower Volume (L) =	1500													
Person Present =	No													

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	unknown	H <sub>c</sub> @ 42 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure
Duration (min) =	12	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)
Liquid Temperature ( C) =	42	1,2-Dibromo-	0.03	7.6 x 10 <sup>-6</sup>	0.056	1.7	1.3	0	n/m	20	1.6	*	*	*
Liquid Flowrate (L/min) =	5.0	3-chloropropane												
Gas Flowrate (L/min) =	53													
Shower Volume (L) =	1500													
Person Present =	No													

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	unknown	H <sub>c</sub> @ 29 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure
Duration (min) =	11	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)
Liquid Temperature ( C) =	29	1,2-Dibromo-	0.012	7.6 x 10 <sup>-6</sup>	0.056	1.5	1.4	0	n/m	5.5	5.5	*	*	*
Liquid Flowrate (L/min) =	10	3-chloropropane												
Gas Flowrate (L/min) =	28													
Shower Volume (L) =	1500													
Person Present =	No													

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	unknown	H <sub>c</sub> @ 32 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure
Duration (min) =	11	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)
Liquid Temperature ( C) =	32	1,2-Dibromo-	0.015	7.6 x 10 <sup>-6</sup>	0.056	2.8	2.4	0	n/m	13	0.97	*	*	*
Liquid Flowrate (L/min) =	5.0	3-chloropropane												
Gas Flowrate (L/min) =	88													
Shower Volume (L) =	1500													
Person Present =	No													

**STUDY:** Giardino and Hageman  
**Study year:** 1996  
**Reference:** *Environmental Science & Technology, Vol. 30, No. 4, 1996, pp.1242-1244.*  
**Solution Method:** Method 1  
**Assumptions:** No  $C_{i,out}$  values were reported, assumed it equal to  $(1-\eta) \times C_{i,in}$ , where  $\eta$  is given in paper.  
**Comments:** The reported value of  $\eta$  is an average value based on pairs of influent and effluent liquid samples taken during experiment.

Liquid-phase diffusion coefficient estimated using relationship 
$$\frac{D_{i,Rn}}{D_{i,Volume}} = \left( \frac{MW_{Volume}}{MW_{Rn}} \right)^{0.5}$$

\* = unable to be determined with available data, n/m = not measured

37		Operating Conditions	Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =		unknown	$H_c @ 22\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{i,in}$	$C_{i,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =		10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)	
Liquid Temperature ( C) =		22	Radon	4.4	$5.2 \times 10^{-6}$	*	$1.1 \times 10^4$	$3.6 \times 10^3$	0	48	67	4.5	*	*	*	51
Liquid Flowrate (L/min) =		4.0														
Gas Flowrate (L/min) =		32														
Shower Volume (L) =		1510														
Person Present =		No														

  

38		Operating Conditions	Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =		unknown	$H_c @ 22\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{i,in}$	$C_{i,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =		10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)	
Liquid Temperature ( C) =		22	Radon	4.4	$5.2 \times 10^{-6}$	*	$3.7 \times 10^4$	$1.1 \times 10^3$	0	17	70	4.9	*	*	*	49
Liquid Flowrate (L/min) =		4.0														
Gas Flowrate (L/min) =		36														
Shower Volume (L) =		1510														
Person Present =		No														

  

39		Operating Conditions	Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =		unknown	$H_c @ 22\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{i,in}$	$C_{i,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =		10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)	
Liquid Temperature ( C) =		22	Radon	4.4	$5.2 \times 10^{-6}$	*	$2.1 \times 10^4$	$8.3 \times 10^2$	0	3.6	61	1.9	*	*	*	53
Liquid Flowrate (L/min) =		2.0														
Gas Flowrate (L/min) =		37														
Shower Volume (L) =		1510														
Person Present =		No														

  

40		Operating Conditions	Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =		unknown	$H_c @ 22\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{i,in}$	$C_{i,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =		10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)	
Liquid Temperature ( C) =		22	Radon	4.4	$5.2 \times 10^{-6}$	*	$3.1 \times 10^4$	$1.1 \times 10^3$	0	9.5	64	2.0	*	*	*	62
Liquid Flowrate (L/min) =		2.0														
Gas Flowrate (L/min) =		39														
Shower Volume (L) =		1510														
Person Present =		No														

**STUDY:** Bernhardt and Hess  
**Study year:** 1995  
**Reference:** Bernhardt's Master's Thesis for The University of Maine  
**Solution Method:** Method 1  
**Assumptions:** Shower stall and bathroom volumes were not given, so assumed a typical bathroom volume of 10000 L.  
 Shower curtain open for all experiments, thus used bathroom volume for  $V_g$ .  
 $Q_g = 0$   
 $C_{g,in} = 0$   
 Exact water temperature not given, so assumed  $T = 23$  C for "cold" water.  
**Comments:** Liquid-phase diffusion coefficient estimated using relationship given for Giardino and Hageman (1996)  
 \* = unable to be determined with available data, n/m = not measured

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	unknown	$H_c @ 23$ C	$D_l @ 24$ C	$D_g @ 24$ C	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure	
Duration (min) =	18	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	23	Radon	4.4	$5.2 \times 10^{-6}$	*	$5.3 \times 10^4$	$1.5 \times 10^4$	0	90	71	2.5	*	*	*	76
Liquid Flowrate (L/min) =	2.0														
Gas Flowrate (L/min) =	0														
Bathroom Volume (L) =	10000														
Person Present =	No														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	unknown	$H_c @ 23$ C	$D_l @ 24$ C	$D_g @ 24$ C	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure	
Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	23	Radon	4.4	$5.2 \times 10^{-6}$	*	$4.9 \times 10^4$	$5.7 \times 10^3$	0	110	88	6.4	*	*	*	87
Liquid Flowrate (L/min) =	3.0														
Gas Flowrate (L/min) =	0														
Bathroom Volume (L) =	10000														
Person Present =	No														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	unknown	$H_c @ 23$ C	$D_l @ 24$ C	$D_g @ 24$ C	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure	
Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	23	Radon	4.35	$5.2 \times 10^{-6}$	*	$4.1 \times 10^4$	$9.0 \times 10^3$	0	300	78	8.3	*	*	*	156
Liquid Flowrate (L/min) =	5.5														
Gas Flowrate (L/min) =	0														
Bathroom Volume (L) =	10000														
Person Present =	No														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Spray Type =	unknown	$H_c @ 23$ C	$D_l @ 24$ C	$D_g @ 24$ C	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure	
Duration (min) =	13	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	23	Radon	4.35	$5.2 \times 10^{-6}$	*	860	280	0	3.8	67	4.2	*	*	*	122
Liquid Flowrate (L/min) =	3.8														
Gas Flowrate (L/min) =	0														
Bathroom Volume (L) =	10000														
Person Present =	No														







**STUDY:** Keating and McKone  
**Study year:** 1993  
**Reference:** *Modeling of Indoor Air Quality and Exposure, ASTM STP 1205, 1993, pp. 14-24*  
**Solution Method:** Method 1  
**Assumptions:**  $C_{g,in} = 0$   
**Comments:**  $C_{g,end}$  values visually picked off graph in paper.

Values for each nozzle represent average of three simulations.

\* = unable to be determined with available data, n/m = not measured

54	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
	Spray Type =	1500 $\mu\text{m}$	$H_c @ 45\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure
	Duration (min) =	8	Chemical	$(\text{m}^3_{\text{liq}}/\text{m}^3_{\text{gas}})$	$(\text{cm}^2/\text{sec})$	$(\text{mg/L})$	$(\text{mg/L})$	$(\text{mg/L})$	$(\text{mg/L})$	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	45	Trichloroethene	0.86	$9.4 \times 10^{-6}$	0.21	0.030	0	0.0050	86	8.8	*	*	*	109
	Liquid Flowrate (L/min) =	4.2													
	Gas Flowrate (L/min) =	65													
	Shower Stall Volume (L) =	1050													
	Person Present =	No													
55	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
	Spray Type =	300 $\mu\text{m}$	$H_c @ 45\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure
	Duration (min) =	10	Chemical	$(\text{m}^3_{\text{liq}}/\text{m}^3_{\text{gas}})$	$(\text{cm}^2/\text{sec})$	$(\text{mg/L})$	$(\text{mg/L})$	$(\text{mg/L})$	$(\text{mg/L})$	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	45	Trichloroethene	0.86	$9.4 \times 10^{-6}$	0.22	0.0063	0	0.0037	97	14	*	*	*	86
	Liquid Flowrate (L/min) =	2.8													
	Gas Flowrate (L/min) =	65													
	Shower Stall Volume (L) =	1050													
	Person Present =	No													
56	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
	Spray Type =	1000 $\mu\text{m}$	$H_c @ 45\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure
	Duration (min) =	5	Chemical	$(\text{m}^3_{\text{liq}}/\text{m}^3_{\text{gas}})$	$(\text{cm}^2/\text{sec})$	$(\text{mg/L})$	$(\text{mg/L})$	$(\text{mg/L})$	$(\text{mg/L})$	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	45	Trichloroethene	0.86	$9.4 \times 10^{-6}$	0.21	0.22	0	0.0032	72	15	*	*	*	72
	Liquid Flowrate (L/min) =	6.0													
	Gas Flowrate (L/min) =	65													
	Shower Stall Volume (L) =	1050													
	Person Present =	No													

**STUDY:** Giardino, Esmen, and Andelman  
**Study year:** 1992  
**Reference:** *Environmental Science and Technology*, Vol. 26, 1992, pp.1602-1606  
**Solution Method:** N/A  
**Assumptions:** No  $C_{i,out}$  values were reported, assumed it equal to  $(1-\eta) \times C_{i,in}$ , where  $\eta$  is given in paper.  
**Comments:** Shower had a vertical spray with no water impacting stall walls.  
 Not enough information given in the paper to calculate  $K_LA$  values or mass closure  
 Gas-phase concentration reported is average value for entire experiment.

\* = unable to be determined with available data, n/m = not measured

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	standard	$H_c @ 22\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{i,in}$	$C_{i,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_LA$	$k_lA$	$k_gA$	$k_g/k_l$	Mass Closure
Duration (min) =	not given	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)
Liquid Temperature ( C) =	22	Trichloroethene	0.37	$9.4 \times 10^{-6}$	0.084	0.46	0.15	0	0.025	67	*	*	*	*
Liquid Flowrate (L/min) =	5													
Gas Flowrate (L/min) =	70													
Shower Stall Volume (L) =	not given													
Person Present =	No													

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	standard	$H_c @ 21\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{i,in}$	$C_{i,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_LA$	$k_lA$	$k_gA$	$k_g/k_l$	Mass Closure
Duration (min) =	not given	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)
Liquid Temperature ( C) =	21	Trichloroethene	0.36	$9.4 \times 10^{-6}$	0.084	0.78	0.32	0	0.080	59	*	*	*	*
Liquid Flowrate (L/min) =	10													
Gas Flowrate (L/min) =	26													
Shower Stall Volume (L) =	not given													
Person Present =	No													

**STUDY:** Tancrede, Yanagisawa, and Wilson  
**Study year:** 1992  
**Reference:** *Atmospheric Environment*, Vol. 26A, No. 6, 1992, pp. 1103-1111  
**Solution Method:** Method 1  
**Assumptions:**  $C_{g,in} = 0$   
**Comments:**  $C_{i,in}$  is average of showerhead liquid samples collected at 2, 6 and 11 minutes.  
 $C_{i,out}$  is average of drain liquid samples collected at 8 and 12 minutes.  
 One  $C_g$  sample was collected at 10 minutes at nose level in the shower stall.

\* = unable to be determined with available data, n/m = not measured

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	unknown	$H_c @ 25\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{i,in}$	$C_{i,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_LA$	$k_lA$	$k_gA$	$k_g/k_l$	Mass Closure
Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)
Liquid Temperature ( C) =	25	$CCl_4$	1.2	$9.2 \times 10^{-6}$	0.072	$8.2 \times 10^{-5}$	$5.0 \times 10^{-5}$	0	$1.4 \times 10^{-6}$	39	7.0	7.2	205	29
Liquid Flowrate (L/min) =	13.6	PCE	0.74	$8.5 \times 10^{-6}$	0.077	$1.1 \times 10^{-4}$	$4.9 \times 10^{-5}$	0	$2.7 \times 10^{-5}$	56	12	13	365	29
Gas Flowrate (L/min) =	34.8	Trichloroethene	0.42	$9.4 \times 10^{-6}$	0.084	0.0016	$6.8 \times 10^{-4}$	0	$4.9 \times 10^{-5}$	58	14	16	446	29
Shower Stall Volume (L) =	1491	Chloroform	0.17	$9.7 \times 10^{-6}$	0.10	0.0026	0.0015	0	$7.8 \times 10^{-5}$	42	11	13	372	29
Person Present =	No	TCPA	0.012	$7.9 \times 10^{-6}$	0.073	0.092	0.073	0	$3.9 \times 10^{-4}$	21	2.5	9.8	281	29





**STUDY:** Jo, Weisel, and Lioy  
**Study year:** 1990  
**Reference:** *Risk Analysis, Vol. 10, No.4, 1992, pp.581-585*  
**Solution Method:** Method 2  
**Assumptions:**  $C_{g,in} = 0$ .  
 $Q_g = 0$ .

**Comments:** Assumed shower air concentration increased linearly such that average gas-phase concentration of entire shower event occurred at the midpoint.  
 Experiments were started two minutes after starting water through shower nozzle.  
 Gas samples are average value of 10 minute sample collection.

\* = unable to be determined with available data, n/m = not measured

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
		$H_c @ 40\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure
		( $m^3_{liq}/m^3_{gas}$ )	( $cm^2/sec$ )	( $cm^2/sec$ )	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
Spray Type =	unknown													
Duration (min) =	10	Chemical							*	2.0	*	*	*	*
Liquid Temperature ( C) =	40	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.013	n/m	0	$6.9 \times 10^{-5}$	*	2.0	*	*	*
Liquid Flowrate (L/min) =	8.7	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.013	n/m	0	$5.8 \times 10^{-5}$	*	1.6	*	*	*
Gas Flowrate (L/min) =	0	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.020	n/m	0	$1.2 \times 10^{-4}$	*	2.4	*	*	*
Shower Stall Volume (L) =	1666	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.021	n/m	0	$9.09 \times 10^{-5}$	*	1.6	*	*	*
Person Present =	No	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.023	n/m	0	$9.0 \times 10^{-5}$	*	1.4	*	*	*
		Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.024	n/m	0	$1.2 \times 10^{-4}$	*	1.8	*	*	*
		Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.024	n/m	0	$2.0 \times 10^{-4}$	*	3.4	*	*	*
		Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.025	n/m	0	$1.7 \times 10^{-4}$	*	2.8	*	*	*
		Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.027	n/m	0	$1.7 \times 10^{-4}$	*	2.4	*	*	*
		Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.028	n/m	0	$2.0 \times 10^{-4}$	*	2.8	*	*	*
		Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.031	n/m	0	$2.0 \times 10^{-4}$	*	2.5	*	*	*
		Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.032	n/m	0	$2.3 \times 10^{-4}$	*	2.8	*	*	*
		Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.040	n/m	0	$3.3 \times 10^{-4}$	*	3.3	*	*	*

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
		$H_c @ 40\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure
		( $m^3_{liq}/m^3_{gas}$ )	( $cm^2/sec$ )	( $cm^2/sec$ )	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
Spray Type =	unknown													
Duration (min) =	10	Chemical							*	2.2	*	*	*	*
Liquid Temperature ( C) =	40	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.022	n/m	0	$1.3 \times 10^{-4}$	*	1.9	*	*	*
Liquid Flowrate (L/min) =	8.7	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.023	n/m	0	$1.2 \times 10^{-4}$	*	2.0	*	*	*
Gas Flowrate (L/min) =	0	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.025	n/m	0	$1.3 \times 10^{-4}$	*	2.6	*	*	*
Shower Stall Volume (L) =	1666	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.029	n/m	0	$2.0 \times 10^{-4}$	*	3.1	*	*	*
Person Present =	Yes	Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.029	n/m	0	$2.3 \times 10^{-4}$	*	3.6	*	*	*
		Chloroform	0.37	$9.7 \times 10^{-6}$	0.10	0.036	n/m	0	$3.1 \times 10^{-4}$	*	3.6	*	*	*

**STUDY:** Giardino, Andelman, Borrazzo, and Davidson  
**Study year:** 1988  
**Reference:** *Journal of the Air Pollution Control Association*, Vol. 38, No. 3, 1988, pp. 278-280  
**Solution Method:** Method 1  
**Assumptions:**  $C_{g,in} = 0$   
 $Q_g = 0$   
**Comments:**  $C_{l,out}$  is average value based on volatilization value reported in paper.  
 \* = unable to be determined with available data, n/m = not measured

70	Operating Conditions		Chemical Properties				Chemical Concentrations				Mass Transfer Parameters					
			$H_c @ 21\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_{LA}$	$k_A$	$k_gA$	$k_g/k_l$	Mass Closure	
		Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)	
	Spray Type =	unknown														
	Duration (min) =	17	SF <sub>6</sub>	185	$1.02 \times 10^{-9}$	*	18	10	0	n/m	44	3.5	*	*	*	
	Liquid Temperature (C) =	21	SF <sub>6</sub>	185	$1.02 \times 10^{-9}$	*	8.8	5.6	0	n/m	36	2.7	*	*	*	
	Liquid Flowrate (L/min) =	6.0	SF <sub>6</sub>	185	$1.02 \times 10^{-9}$	*	18	8.5	0	n/m	52	4.4	*	*	*	
	Gas Flowrate (L/min) =	0	SF <sub>6</sub>	185	$1.02 \times 10^{-9}$	*	26	13	0	n/m	51	4.3	*	*	*	
	Shower Stall Volume (L) =	1100	SF <sub>6</sub>	185	$1.02 \times 10^{-9}$	*	24	12	0	n/m	48	3.9	*	*	*	
	Person Present =	No	SF <sub>6</sub>	185	$1.02 \times 10^{-9}$	*	15	9.2	0	n/m	39	3.0	*	*	*	
			SF <sub>6</sub>	185	$1.02 \times 10^{-9}$	*	19	9.4	0	n/m	50	4.2	*	*	*	
			SF <sub>6</sub>	185	$1.02 \times 10^{-9}$	*	20	11	0	n/m	45	3.6	*	*	*	

**STUDY:** Hodgson, Garbesi, Sextro, and Daisey  
**Study year:** 1988  
**Reference:** Lawrence Berkeley Laboratory Report, Contract No. DE-Ac03-76SF00098, 1988  
**Solution Method:** Method 1  
**Assumptions:** Assumed entire bathroom including shower stall was well-mixed, such that only bathroom measurements were used to predict  $K_{LA}$ .  
**Comments:** For Freon-12, used Henry's law constant given in paper, remaining chemical Henry's law constants using Ashworth *et al.* correlations.  
 \* = unable to be determined with available data, n/m = not measured

71	Operating Conditions		Chemical Properties				Chemical Concentrations				Mass Transfer Parameters					
			$H_c @ 40\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_{LA}$	$k_A$	$k_gA$	$k_g/k_l$	Mass Closure	
		Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)	
	Spray Type =	unknown														
	Duration (min) =	10	Freon - 12	32	$1.0 \times 10^{-4}$	*	0.073	0.010	$4.2 \times 10^{-6}$	$4.3 \times 10^{-5}$	86	27	*	*	23	
	Liquid Temperature (C) =	40	Freon - 11	6.8	$9.0 \times 10^{-6}$	0.084	0.011	0.0010	$4.0 \times 10^{-6}$	$1.1 \times 10^{-5}$	91	33	34	149	4.4	
	Liquid Flowrate (L/min) =	13.7	PCE	1.4	$8.5 \times 10^{-6}$	0.077	0.018	0.0032	$6.0 \times 10^{-6}$	$2.5 \times 10^{-5}$	82	24	28	122	4.4	
	Gas Flowrate (L/min) =	51	TCA	1.2	$9.0 \times 10^{-6}$	0.080	0.0034	$5.0 \times 10^{-4}$	$1.4 \times 10^{-6}$	$6.6 \times 10^{-5}$	85	27	32	140	4.4	
	Bathroom Volume (L) =	10900	Trichloroethene	0.73	$9.4 \times 10^{-6}$	0.084	0.0027	$4.0 \times 10^{-4}$	0	n/m	85	27	35	154	4.4	
	Person Present =	No													*	

72	Operating Conditions		Chemical Properties				Chemical Concentrations				Mass Transfer Parameters					
			$H_c @ 40\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_{LA}$	$k_A$	$k_gA$	$k_g/k_l$	Mass Closure	
		Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)	
	Spray Type =	unknown														
	Duration (min) =	10	Freon - 12	32	$1.0 \times 10^{-4}$	*	0.094	0.018	$6.6 \times 10^{-6}$	$6.7 \times 10^{-5}$	81	23	*	*	30	
	Liquid Temperature (C) =	40	Freon - 11	6.8	$9.0 \times 10^{-6}$	0.084	0.0094	0.0018	$4.5 \times 10^{-6}$	$1.2 \times 10^{-5}$	81	23	*	*	37	
	Liquid Flowrate (L/min) =	13.7	PCE	1.4	$8.5 \times 10^{-6}$	0.077	0.029	0.0042	$3.6 \times 10^{-6}$	$2.2 \times 10^{-5}$	86	27	*	*	26	
	Gas Flowrate (L/min) =	51	TCA	1.2	$9.0 \times 10^{-6}$	0.080	0.0031	$8.0 \times 10^{-4}$	$6.6 \times 10^{-6}$	$9.6 \times 10^{-6}$	74	19	*	*	57	
	Bathroom Volume (L) =	10900	Trichloroethene	0.73	$9.4 \times 10^{-6}$	0.084	0.0030	$4.0 \times 10^{-4}$	0	n/m	87	28	*	*	*	
	Person Present =	Yes													*	



**STUDY:** Hess *et al.*  
**Study year:** 1982  
**Reference:** *Environment International, Vol. 8, 1982, pp. 59-66.*  
**Solution Method:** N/A  
**Assumptions:** Stripping efficiency is value reported in paper. No data was given to confirm.  
**Comments:** Stripping efficiency based on four measurements.

Not enough data collected to calculate  $K_LA$  or mass closure.

\* = unable to be determined with available data, n/m = not measured

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Operating Conditions		Chemical Properties				Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	unknown	$H_c @ 25\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_LA$	$k_1A$	$k_gA$	$k_g/k_l$	Mass Closure	
Duration (min) =	not given	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)	
Liquid Temperature ( C) =	n/m	Radon	4.4	$5.2 \times 10^{-6}$	*	*	*	*	65	*	*	*	*	*	
Liquid Flowrate (L/min) =	n/m														
Gas Flowrate (L/min) =	n/m														
Bathroom Volume (L) =	n/m														
Person Present =	No														

**STUDY:** Gesell and Prichard  
**Study year:** 1980  
**Reference:** In *Natural Radiation Environment III, Vol. 2*, Houston: Technical Information Center, U.S. Department of Energy, pp. 1347-1363.  
**Solution Method:** N/A  
**Assumptions:** Stripping efficiency is value reported in paper. No data was given to confirm.  
**Comments:** Liquid-phase diffusion coefficient estimated using relationship given for Giardino and Hageman (1996).  
 Not enough data collected to calculate  $K_LA$  or mass closure.

\* = unable to be determined with available data, n/m = not measured

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Operating Conditions		Chemical Properties				Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	unknown	$H_c @ 25\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_LA$	$k_1A$	$k_gA$	$k_g/k_l$	Mass Closure	
Duration (min) =	not given	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)	
Liquid Temperature ( C) =	n/m	Radon	4.4	$5.2 \times 10^{-6}$	*	*	*	*	65	*	*	*	*	*	
Liquid Flowrate (L/min) =	n/m														
Gas Flowrate (L/min) =	n/m														
Bathroom Volume (L) =	n/m														
Person Present =	No														

**STUDY:** Partridge *et al.*  
**Study year:** 1979  
**Reference:** Data in Nazaroff *et al.*, *Health Physics*, Vol. 52, No. 3, 1987, pp. 281-295.  
**Solution Method:** N/A  
**Assumptions:** Stripping efficiency is value reported in paper. No data was given to confirm.  
**Comments:** Liquid-phase diffusion coefficient estimated using relationship given for Giardino and Hageman (1996)  
 Not enough data collected to calculate  $K_LA$  or mass closure.

\* = unable to be determined with available data, n/m = not measured

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Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
Spray Type =	unknown	$H_c @ 25 C$	$D_l @ 24 C$	$D_g @ 24 C$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_LA$	$k_lA$	$k_gA$	$k_g/k_l$	Mass Closure
Duration (min) =	not given	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	$(mg/L)$	(%)	(L/min)	(L/min)	(L/min)		(%)
Liquid Temperature (C) =	n/m	Radon	4.4	$5.2 \times 10^{-6}$	*	*	*	*	71	*	*	*	*	*
Liquid Flowrate (L/min) =	n/m													
Gas Flowrate (L/min) =	n/m													
Bathroom Volume (L) =	n/m													
Person Present =	No													

**BATHTUB DATABASE**

**STUDY:** Howard and Corsi  
**Study year:** 1997  
**Solution Methods:** Method 1 for ethyl acetate, toluene, ethylbenzene, cyclohexane  
 Method 2 for acetone  
**Assumptions:** None  
**Comments:** Values of  $\eta$  and  $K_L A$  are averages of values determined for three separate time periods within experiment  
 \* = unable to be determined with available data

Entry #	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
1	Operation =	Flow-through	$H_c @ 22\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_i A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
	Liquid Temperature ( C) =	22	Acetone	0.0011	$1.1 \times 10^{-5}$	0.11	52	50	0	0.019	3.8	0.11	2.9	108	37	98
	Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0044	$9.5 \times 10^{-6}$	0.092	25	24	0	0.028	6.1	0.64	4.5	168	37	98
	Gas Flowrate (L/min) =	355	Toluene	0.25	$9.1 \times 10^{-6}$	0.085	4.9	3.6	0	0.019	26	2.9	3.2	117	37	89
	Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.28	$8.4 \times 10^{-6}$	0.077	3.9	2.9	0	0.012	27	2.9	3.1	117	37	86
	Person Present =	No	Cyclohexane	6.7	$9.0 \times 10^{-6}$	0.088	0.94	0.68	0	0.0031	28	3.1	3.1	115	37	85
2	Operation =	Flow-through	$H_c @ 23\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_i A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
	Liquid Temperature ( C) =	23	Acetone	0.0011	$1.1 \times 10^{-5}$	0.11	40	39	0	0.016	3.1	0.15	3.2	136	43	99
	Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0046	$9.5 \times 10^{-6}$	0.092	25	23	0	0.027	4.7	0.49	3.0	126	43	102
	Gas Flowrate (L/min) =	345	Toluene	0.26	$9.1 \times 10^{-6}$	0.085	6.4	4.9	0	0.019	24	2.4	2.6	111	43	92
	Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.30	$8.4 \times 10^{-6}$	0.077	4.9	3.8	0	0.0088	24	2.4	2.6	109	43	88
	Person Present =	No	Cyclohexane	6.9	$9.0 \times 10^{-6}$	0.088	1.4	1.0	0	0.0032	29	2.9	2.9	126	43	88
3	Operation =	Flow-through	$H_c @ 36\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_i A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
	Liquid Temperature ( C) =	36	Acetone	0.0024	$1.1 \times 10^{-5}$	0.11	39	37	0	0.036	5.3	0.54	6.0	249	42	99
	Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0080	$9.5 \times 10^{-6}$	0.092	24	21	0	0.059	11	1.2	4.9	205	42	102
	Gas Flowrate (L/min) =	359	Toluene	0.38	$9.1 \times 10^{-6}$	0.085	6.7	4.1	0	0.037	38	4.5	4.8	198	42	94
	Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.58	$8.4 \times 10^{-6}$	0.077	5.9	3.6	0	0.022	39	4.5	4.7	196	42	87
	Person Present =	No	Cyclohexane	10	$9.0 \times 10^{-6}$	0.088	2.3	1.3	0	0.0074	38	5.1	5.1	211	42	82
4	Operation =	Flow-through	$H_c @ 25\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_i A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
	Liquid Temperature ( C) =	25	Acetone	0.0013	$1.1 \times 10^{-5}$	0.11	40	40	0	0.017	1.7	0.18	2.4	159	66	102
	Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0050	$9.5 \times 10^{-6}$	0.092	25	24	0	0.029	4.5	0.32	1.3	86	66	103
	Gas Flowrate (L/min) =	350	Toluene	0.27	$9.1 \times 10^{-6}$	0.085	5.2	4.0	0	0.013	22	1.6	1.7	110	66	95
	Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.32	$8.4 \times 10^{-6}$	0.077	4.0	3.1	0	0.0077	22	1.5	1.6	106	66	92
	Person Present =	No	Cyclohexane	7.2	$9.0 \times 10^{-6}$	0.088	1.1	0.81	0	0.0029	22	1.7	1.7	110	66	96

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Flow-through	$H_c @ 36\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	36	Acetone	0.0024	$1.1 \times 10^{-5}$	0.11	48	46	0	0.034	4.3	0.43	2.4	227	96	101
Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0080	$9.5 \times 10^{-6}$	0.092	25	22	0	0.043	14	1.1	2.4	234	96	101
Gas Flowrate (L/min) =	361	Toluene	0.38	$9.1 \times 10^{-6}$	0.085	6.0	4.2	0	0.023	30	2.2	2.2	214	96	107
Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.58	$8.4 \times 10^{-6}$	0.077	4.0	3.1	0	0.017	29	2.1	2.2	207	96	100
Person Present =	No	Cyclohexane	10	$9.0 \times 10^{-6}$	0.088	1.1	0.77	0	0.0038	27	1.9	1.9	182	96	103

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Flow-through	$H_c @ 37\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	37	Acetone	0.0025	$1.1 \times 10^{-5}$	0.11	40	38	0	0.033	4.8	0.46	4.2	211	50	101
Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0084	$9.5 \times 10^{-6}$	0.092	23	21	0	0.049	10	0.79	2.7	135	50	105
Gas Flowrate (L/min) =	365	Toluene	0.39	$9.1 \times 10^{-6}$	0.085	6.0	3.7	0	0.023	38	2.9	3.1	152	50	92
Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.59	$8.4 \times 10^{-6}$	0.077	5.9	3.7	0	0.015	38	2.9	3.0	151	50	85
Person Present =	No	Cyclohexane	11	$9.0 \times 10^{-6}$	0.088	2.0	1.2	0	0.0038	41	3.2	3.2	161	50	85

STUDY: Howard and Corsi  
Study year: 1997  
Solution Methods: Method 3  
Assumptions: None  
Comments:  
\* = unable to be determined with available data

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Fill	$H_c @ 24\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	24	Acetone	0.0012	$1.1 \times 10^{-5}$	0.11	41	39	0	0.018	4.9	0.45	7.1	395	56	97
Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0013	$9.5 \times 10^{-6}$	0.092	25	24	0	0.030	3.0	1.0	4.9	274	56	103
Final Liquid Volume (L) =	73	Toluene	0.27	$9.1 \times 10^{-6}$	0.085	8.0	5.6	0	0.029	31	4.1	4.4	244	56	89
Gas Flowrate (L/min) =	373	Ethylbenzene	0.31	$8.4 \times 10^{-6}$	0.077	9.8	6.7	0	0.020	33	4.4	4.6	257	56	82
Shower Stall/Bathtub Volume (L) =	1745	Cyclohexane	7.1	$9.0 \times 10^{-6}$	0.088	3.1	1.8	0	0.0063	46	7.1	7.1	396	56	73

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Fill	$H_c @ 35\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
Duration (min) =	8	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	35	Acetone	0.0022	$1.1 \times 10^{-5}$	0.11	43	40	0	0.031	5.2	0.53	9.3	253	27	98
Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0077	$9.5 \times 10^{-6}$	0.092	24	23	0	0.045	5.3	1.4	8.3	228	27	106
Final Liquid Volume (L) =	73	Toluene	0.36	$9.1 \times 10^{-6}$	0.085	6.2	4.0	0	0.036	30	5.3	5.8	159	27	93
Gas Flowrate (L/min) =	379	Ethylbenzene	0.54	$8.4 \times 10^{-6}$	0.077	6.3	3.9	0	0.024	32	5.9	6.3	172	27	81
Shower Stall/Bathtub Volume (L) =	1745	Cyclohexane	10	$9.0 \times 10^{-6}$	0.088	2.0	0.85	0	0.0074	47	11	11	311	27	68

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Fill	H <sub>c</sub> @ 36 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	8	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	36	Acetone	0.0024	1.1 x 10 <sup>-5</sup>	0.11	40	39	0	0.032	2.0	0.64	5.9	303	51	101
Liquid Flowrate (L/min) =	9.1	Ethyl Acetate	0.0080	9.5 x 10 <sup>-6</sup>	0.092	23	22	0	0.044	3.1	1.5	5.3	269	51	107
Final Liquid Volume (L) =	73	Toluene	0.38	9.1 x 10 <sup>-6</sup>	0.085	6.3	4.5	0	0.037	31	3.7	3.8	193	51	101
Gas Flowrate (L/min) =	373	Ethylbenzene	0.58	8.4 x 10 <sup>-6</sup>	0.077	5.4	3.8	0	0.024	32	3.8	4.0	202	51	93
Shower Stall/Bathtub Volume (L) =	1745	Cyclohexane	10	9.0 x 10 <sup>-6</sup>	0.088	1.2	0.66	0	0.0074	46	7.4	7.4	376	51	87

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Fill	H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	12	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	23	Acetone	0.0011	1.1 x 10 <sup>-5</sup>	0.11	41	39	0	0.019	5.8	0.39	4.7	365	77	96
Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0046	9.5 x 10 <sup>-6</sup>	0.092	26	25	0	0.032	3.1	0.71	2.7	208	77	104
Final Liquid Volume (L) =	73	Toluene	0.26	9.1 x 10 <sup>-6</sup>	0.085	6.8	4.9	0	0.034	29	2.6	2.7	208	77	101
Gas Flowrate (L/min) =	370	Ethylbenzene	0.30	8.4 x 10 <sup>-6</sup>	0.077	7.3	5.1	0	0.023	31	2.7	2.8	220	77	88
Shower Stall/Bathtub Volume (L) =	1745	Cyclohexane	6.9	9.0 x 10 <sup>-6</sup>	0.088	3.0	1.8	0	0.0078	43	4.4	4.4	344	77	84

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Fill	H <sub>c</sub> @ 35 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	11.4	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	35	Acetone	0.0022	1.1 x 10 <sup>-5</sup>	0.11	40	37	0	0.032	7.7	0.39	4.2	191	46	98
Liquid Flowrate (L/min) =	6.1	Ethyl Acetate	0.0077	9.5 x 10 <sup>-6</sup>	0.092	23	22	0	0.047	7.0	1.0	3.8	175	46	108
Final Liquid Volume (L) =	69	Toluene	0.37	9.1 x 10 <sup>-6</sup>	0.085	5.0	3.5	0	0.026	30	2.7	2.8	129	46	106
Gas Flowrate (L/min) =	377	Ethylbenzene	0.56	8.4 x 10 <sup>-6</sup>	0.077	4.2	3.0	0	0.015	29	2.5	2.6	121	46	96
Shower Stall/Bathtub Volume (L) =	1745	Cyclohexane	10	9.0 x 10 <sup>-6</sup>	0.088	1.2	0.63	0	0.0041	46	5.4	5.4	245	46	75

**STUDY:** Howard and Corsi  
**Study year:** 1997  
**Solution Methods:** Method 4 for toluene, ethylbenzene, and cyclohexane  
Method 5 for acetone and ethyl acetate  
**Assumptions:** None  
**Comments:** Stripping efficiencies for entries 12 - 14 based on gas-phase data  
\* = unable to be determined with available data

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Bathing	H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	20	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	23	Acetone	0.0011	1.1 x 10 <sup>-5</sup>	0.11	40	39	0.021	0.0030	0.6	*	*	*	*	99
Liquid Volume (L) =	73	Ethyl Acetate	0.0046	9.5 x 10 <sup>-6</sup>	0.092	25	25	0.035	0.0023	1.6	*	*	*	*	100
Gas Flowrate (L/min) =	370	Toluene	0.26	9.1 x 10 <sup>-6</sup>	0.085	4.9	5.1	0.037	3.8 x 10 <sup>-1</sup>	7.9	*	*	*	*	96
Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.3	8.4 x 10 <sup>-6</sup>	0.077	5.1	5.3	0.025	4.2 x 10 <sup>-1</sup>	5.1	*	*	*	*	100
Person Present =	No	Cyclohexane	6.9	9.0 x 10 <sup>-6</sup>	0.088	1.8	1.9	0.009	0	4.7	*	*	*	*	99

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Bathing	$H_c @ 34\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure	
Duration (min) =	20	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	34	Acetone	0.0012	$1.1 \times 10^{-5}$	0.11	39	39	0.016	0.0033	2.5	*	*	*	*	101
Liquid Volume (L) =	69	Ethyl Acetate	0.0048	$9.5 \times 10^{-6}$	0.092	24	24	0.027	0.0034	5.9	*	*	*	*	102
Gas Flowrate (L/min) =	377	Toluene	0.26	$9.1 \times 10^{-6}$	0.085	3.3	3.5	0.021	$9.2 \times 10^{-1}$	13	*	*	*	*	110
Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.31	$8.4 \times 10^{-6}$	0.077	2.9	2.7	0.013	$5.1 \times 10^{-1}$	7.6	*	*	*	*	96
Person Present =	No	Cyclohexane	7.1	$9.0 \times 10^{-6}$	0.088	0.65	0.74	0.0040	$1.4 \times 10^{-1}$	13	*	*	*	*	117

  

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Bathing	$H_c @ 34\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure	
Duration (min) =	20	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	34	Acetone	0.0021	$1.1 \times 10^{-5}$	0.11	37	38	0.032	0.012	2.7	*	*	*	*	104
Final Liquid Volume (L) =	73	Ethyl Acetate	0.0074	$9.5 \times 10^{-6}$	0.092	22	22	0.047	0.0091	6.4	*	*	*	*	105
Gas Flowrate (L/min) =	377	Toluene	0.36	$9.1 \times 10^{-6}$	0.085	3.6	3.6	0.026	0.0016	14	*	*	*	*	99
Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.52	$8.4 \times 10^{-6}$	0.077	3.1	2.7	0.015	0.0010	8.3	*	*	*	*	86
Person Present =	No	Cyclohexane	9.8	$9.0 \times 10^{-6}$	0.088	0.64	0.59	0.004	$2.4 \times 10^{-1}$	15	*	*	*	*	93

  

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Bathing	$H_c @ 24\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure	
Duration (min) =	20	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	24	Acetone	0.0012	$1.1 \times 10^{-5}$	0.11	39	38	0.007	0.011	1.6	0.11	1.8	97	54	103
Final Liquid Volume (L) =	73	Ethyl Acetate	0.0048	$9.5 \times 10^{-6}$	0.092	24	23	0.009	0.016	3.4	0.24	1.2	63	54	109
Gas Flowrate (L/min) =	373	Toluene	0.26	$9.1 \times 10^{-6}$	0.085	5.7	3.9	0.009	0.0092	32	1.2	1.3	69	54	90
Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.31	$8.4 \times 10^{-6}$	0.077	6.9	4.7	0.006	0.0059	32	1.2	1.3	68	54	83
Person Present =	Yes	Cyclohexane	7.1	$9.0 \times 10^{-6}$	0.088	1.7	1.0	0.002	0.0014	39	1.4	1.4	76	54	82

  

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Bathing	$H_c @ 33\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure	
Duration (min) =	20	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	33	Acetone	0.0020	$1.1 \times 10^{-5}$	0.11	41	39	0.012	0.023	4.5	0.25	1.8	143	35	100
Final Liquid Volume (L) =	73	Ethyl Acetate	0.0071	$9.5 \times 10^{-6}$	0.092	23	21	0.013	0.027	9.8	0.49	1.4	107	35	104
Gas Flowrate (L/min) =	379	Toluene	0.35	$9.1 \times 10^{-6}$	0.085	4.2	3.1	0.007	0.0091	27	1.2	1.2	97	35	100
Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.50	$8.4 \times 10^{-6}$	0.077	4.1	3.0	0.005	0.0063	26	1.1	1.1	88	35	92
Person Present =	Yes	Cyclohexane	9.6	$9.0 \times 10^{-6}$	0.088	1.0	0.59	0.002	0.0018	41	2.0	2.0	156	35	80

  

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Operation =	Bathing	$H_c @ 35\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure	
Duration (min) =	20	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	35	Acetone	0.0022	$1.1 \times 10^{-5}$	0.11	41	38	0.01	0.026	7.3	0.23	1.6	122	78	98
Final Liquid Volume (L) =	73	Ethyl Acetate	0.0077	$9.5 \times 10^{-6}$	0.092	22	20	0.008	0.028	8.9	0.40	1.1	84	78	105
Gas Flowrate (L/min) =	373	Toluene	0.36	$9.1 \times 10^{-6}$	0.085	4.7	3.3	0.005	0.011	30	1.2	1.2	97	78	96
Shower Stall/Bathtub Volume (L) =	1745	Ethylbenzene	0.54	$8.4 \times 10^{-6}$	0.077	4.0	2.9	0.003	0.0061	29	1.1	1.1	88	78	88
Person Present =	Yes	Cyclohexane	10	$9.0 \times 10^{-6}$	0.088	0.73	0.51	$1.8 \times 10^{-1}$	0.0016	30	1.2	1.2	94	78	91

**KITCHEN SINK DATABASE**



**STUDY:** Howard and Corsi  
**Study year:** 1996  
**Reference:** *Journal of the Air and Waste Management Association, Vol. 46, 1996, pp. 830-837.*  
**Solution Method:** Method 6  
**Assumptions:**  $C_g$  assumed to be negligible  
 $C_{g,in} = 0$   
**Comments:** Mass transfer values based on recirculating flow.  
 \* = unable to be determined with available data

Entry #	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters					
			$H_c$ @ 23 C	$D_l$ @ 24 C	$D_g$ @ 24 C	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_f$	Mass Closure
			( $m^3_{liq}/m^3_{gas}$ )	( $cm^2/sec$ )	( $cm^2/sec$ )	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
1	Aerator Type =	None	Chemical	Acetone	$1.1 \times 10^{-5}$	93	45	0	0	4.9	0.24	1.3	136	104	*
	Liquid Temperature ( C) =	23	Chemical	Toluene	$9.1 \times 10^{-6}$	11	0.44	0	0	21	1.0	1.0	108	104	*
	Liquid Flowrate (L/min) =	4.8	Chemical	Cyclohexane	$9.0 \times 10^{-6}$	7.2	0.18	0	0	24	1.2	1.2	125	104	*
	Hydraulic Residence Time (min) =	10													
	Dishes =	None													
2	Aerator Type =	None	Chemical	Acetone	$1.1 \times 10^{-5}$	56	41	0	0	2.2	0.17	1.7	88	51	*
	Liquid Temperature ( C) =	23	Chemical	Toluene	$9.1 \times 10^{-6}$	14	1.2	0	0	17	1.3	1.4	72	51	*
	Liquid Flowrate (L/min) =	7.9	Chemical	Cyclohexane	$9.0 \times 10^{-6}$	12	0.74	0	0	19	1.5	1.5	77	51	*
	Hydraulic Residence Time (min) =	6.3													
	Dishes =	None													
3	Aerator Type =	Screen	Chemical	Acetone	$1.1 \times 10^{-5}$	70	53	0	0	1.7	0.080	1.0	41	43	*
	Liquid Temperature ( C) =	23	Chemical	Toluene	$9.1 \times 10^{-6}$	9.4	0.84	0	0	13	0.65	0.7	30	43	*
	Liquid Flowrate (L/min) =	4.8	Chemical	Cyclohexane	$9.0 \times 10^{-6}$	7.7	0.30	0	0	19	0.90	0.9	38	43	*
	Hydraulic Residence Time (min) =	10													
	Dishes =	None													
4	Aerator Type =	Screen	Chemical	Acetone	$1.1 \times 10^{-5}$	58	49	0	0	1.1	0.090	1.6	44	27	*
	Liquid Temperature ( C) =	23	Chemical	Toluene	$9.1 \times 10^{-6}$	12	1.5	0	0	14	1.2	1.4	37	27	*
	Liquid Flowrate (L/min) =	7.9	Chemical	Cyclohexane	$9.0 \times 10^{-6}$	8.4	0.64	0	0	18	1.4	1.4	38	27	*
	Hydraulic Residence Time (min) =	6.3													
	Dishes =	None													

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Aerator Type =	Bubble Aerator	H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>i,in</sub>	C <sub>i,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>f</sub>	Mass Closure	
Liquid Temperature ( C ) =	23	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Flowrate (L/min) =	4.8	Acetone	0.0011	1.1 x 10 <sup>-5</sup>	0.11	54	42	0	0	1.4	0.065	1.8	31	18	*
Hydraulic Residence Time (min) =	10	Toluene	0.26	9.1 x 10 <sup>-6</sup>	0.085	6.5	0.11	0	0	23	1.1	1.3	24	18	*
Dishes =	None	Cyclohexane	6.9	9.0 x 10 <sup>-6</sup>	0.088	4.2	0.010	0	0	33	1.6	1.6	28	18	*

  

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Aerator Type =	Bubble Aerator	H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>i,in</sub>	C <sub>i,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>f</sub>	Mass Closure	
Liquid Temperature ( C ) =	23	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Flowrate (L/min) =	6.3	Acetone	0.0011	1.1 x 10 <sup>-5</sup>	0.11	56	45	0	0	1.5	0.090	2.4	43	18	*
Hydraulic Residence Time (min) =	8.1	Toluene	0.26	9.1 x 10 <sup>-6</sup>	0.085	7.0	0.2	0	0	22	1.4	1.7	31	18	*
Dishes =	None	Cyclohexane	6.9	9.0 x 10 <sup>-6</sup>	0.088	4.3	0.013	0	0	35	2.2	2.2	41	18	*

  

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Aerator Type =	Bubble Aerator	H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>i,in</sub>	C <sub>i,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>f</sub>	Mass Closure	
Liquid Temperature ( C ) =	23	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Flowrate (L/min) =	7.9	Acetone	0.0011	1.1 x 10 <sup>-5</sup>	0.11	58	54	0	0	1.6	0.13	3.6	63	18	*
Hydraulic Residence Time (min) =	6.3	Toluene	0.26	9.1 x 10 <sup>-6</sup>	0.085	9.8	2.8	0	0	23	1.9	2.3	41	18	*
Dishes =	None	Cyclohexane	6.9	9.0 x 10 <sup>-6</sup>	0.088	7.2	0.70	0	0	44	3.5	3.5	62	18	*

  

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Aerator Type =	Bubble Aerator	H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>i,in</sub>	C <sub>i,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>f</sub>	Mass Closure	
Liquid Temperature ( C ) =	23	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Flowrate (L/min) =	4.8	Acetone	0.0011	1.1 x 10 <sup>-5</sup>	0.11	68	57	0	0	1.3	0.060	2.1	29	14	*
Hydraulic Residence Time (min) =	10	Toluene	0.26	9.1 x 10 <sup>-6</sup>	0.085	5.0	2.8	0	0	24	1.2	1.5	21	14	*
Dishes =	Yes	Cyclohexane	6.9	9.0 x 10 <sup>-6</sup>	0.088	3.8	0.015	0	0	40	1.9	1.9	26	14	*

  

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Aerator Type =	Bubble Aerator	H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>i,in</sub>	C <sub>i,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>f</sub>	Mass Closure	
Liquid Temperature ( C ) =	23	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Flowrate (L/min) =	6.3	Acetone	0.0011	1.1 x 10 <sup>-5</sup>	0.11	81	59	0	0	3.4	0.21	2.8	106	38	*
Hydraulic Residence Time (min) =	8.1	Toluene	0.26	9.1 x 10 <sup>-6</sup>	0.085	8.1	0.44	0	0	26	1.6	1.8	68	38	*
Dishes =	Yes	Cyclohexane	6.9	9.0 x 10 <sup>-6</sup>	0.088	4.2	0.020	0	0	48	2.9	2.9	112	38	*

**STUDY:** Wooley and Nazaroff  
**Study year:** 1990  
**Reference:** *Journal of the Air and Waste Management Association, Vol. 40, 1990, pp.1114-11120*  
**Solution Method:** Method 8  
**Assumptions:** Only significant mass transfer occurring due to wash solution with added ethanol.  
**Comments:** There is not enough information given to consider additional emissions due to rinsing of dishes.  
 Mass closure values based on numbers given in paper.

\* = unable to be determined with available data

10	Operating Conditions		Chemical Properties				Chemical Concentrations				Mass Transfer Parameters				
			H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure
			(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
	Aerator Type =	unknown													
	Liquid Temperature ( C) =	43	Chemical												
	Liquid Volume (L) =	7.6	Ethanol	0.0012	1.3 x 10 <sup>-5</sup>	0.12	113	101	0	0	11	0.084	*	*	94
	Duration (min) =	10	Ethanol	0.0012	1.3 x 10 <sup>-5</sup>	0.12	113	105	0	0	7.1	0.055	*	*	96
	Dishes =	Yes	Ethanol	0.0012	1.3 x 10 <sup>-5</sup>	0.12	113	99	0	0	12	0.10	*	*	92

11	Operating Conditions		Chemical Properties				Chemical Concentrations				Mass Transfer Parameters				
			H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure
			(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)		(%)
	Aerator Type =	unknown													
	Liquid Temperature ( C) =	55	Chemical												
	Liquid Volume (L) =	7.6	Ethanol	*	1.3 x 10 <sup>-5</sup>	0.12	278	279	0	0	-0.4	-5.1 x 10 <sup>-4</sup>	*	*	106
	Duration (min) =	70	Ethanol	*	1.3 x 10 <sup>-5</sup>	0.12	278	283	0	0	-1.8	-0.0020	*	*	106
	Dishes =	Yes	Ethanol	*	1.3 x 10 <sup>-5</sup>	0.12	278	278	0	0	0	0	*	*	105

**WASHING MACHINE DATABASE**

**STUDY:** Howard and Corsi  
**Study year:** 1997  
**Solution Method:** Method 3  
**Assumptions:** None  
**Comments:** One gas sample collected for duration of experiment and one gas sample collected at end of experiment.  
 Values of  $k_g/k_l$  based solely on toluene, ethylbenzene, and cyclohexane data.

\* = unable to be determined with available data

Entry #	Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
1	Cycle Type =	fill for rinse	$H_c @ 19\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	3.3	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(L/min)	(%)
	Liquid Temperature ( C) =	19	Acetone	0.00089	$1.1 \times 10^{-5}$	0.11	55	55	0	0.042	2.1	0.23	*	*	*	99
	Liquid Flowrate (L/min) =	14.6	Toluene	0.22	$9.1 \times 10^{-6}$	0.085	7.5	6.8	0	0.28	9.8	1.8	2.9	21	7.1	108
	Gas Flowrate (L/min) =	55	Ethylbenzene	0.23	$8.4 \times 10^{-6}$	0.077	6.7	6.0	0	0.18	9.5	1.7	2.8	20	7.1	102
	Final Headspace Volume (L) =	101	Cyclohexane	6.0	$9.0 \times 10^{-6}$	0.088	2.5	2.1	0	0.074	15	2.8	2.9	21	7.1	98
Clothes Present =	No															
2	Cycle Type =	fill for rinse	$H_c @ 21\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	3.3	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(L/min)	(%)
	Liquid Temperature ( C) =	21	Acetone	0.0010	$1.1 \times 10^{-5}$	0.11	55	55	0	0.067	1	*	*	*	*	100
	Liquid Flowrate (L/min) =	13.7	Toluene	0.24	$9.1 \times 10^{-6}$	0.085	8.5	7.3	0	0.32	13	2.8	5.4	25	4.5	104
	Gas Flowrate (L/min) =	55	Ethylbenzene	0.27	$8.4 \times 10^{-6}$	0.077	7.3	6.3	0	0.20	13	2.9	5.3	24	4.5	99
	Final Headspace Volume (L) =	104	Cyclohexane	6.0	$9.0 \times 10^{-6}$	0.088	2.1	1.6	0	0.068	25	5.3	5.5	25	4.5	90
Clothes Present =	no															
3	Cycle Type =	fill for wash	$H_c @ 19\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	3.3	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(L/min)	(%)
	Liquid Temperature ( C) =	19	Acetone	0.00089	$1.1 \times 10^{-5}$	0.11	53	53	0	0.061	0.7	*	*	*	*	100
	Liquid Flowrate (L/min) =	13.8	Toluene	0.23	$9.1 \times 10^{-6}$	0.085	7.9	6.3	0	0.25	13	4.2	7.0	47	6.7	96
	Gas Flowrate (L/min) =	55	Ethylbenzene	0.24	$8.4 \times 10^{-6}$	0.077	7.3	5.7	0	0.13	16	5.0	8.1	54	6.7	87
	Final Headspace Volume (L) =	104	Cyclohexane	6.0	$9.0 \times 10^{-6}$	0.088	2.3	1.5	0	0.066	26	7.5	7.6	51	6.7	79
Clothes Present =	no															
4	Cycle Type =	fill for rinse	$H_c @ 21\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_l A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	3.3	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(L/min)	(%)
	Liquid Temperature ( C) =	21	Acetone	0.0010	$1.1 \times 10^{-5}$	0.11	56	54	0	0.038	3.0	0.086	*	*	*	97
	Liquid Flowrate (L/min) =	13.7	Toluene	0.27	$9.1 \times 10^{-6}$	0.085	8.6	7.8	0	0.14	8.2	1.5	*	*	*	102
	Gas Flowrate (L/min) =	55	Ethylbenzene	0.35	$8.4 \times 10^{-6}$	0.077	6.9	6.2	0	0.076	10	1.9	*	*	*	96
	Final Headspace Volume (L) =	93	Cyclohexane	6.5	$9.0 \times 10^{-6}$	0.088	2.0	1.9	0	0.028	6.9	1.2	*	*	*	102
Clothes Present =	yes															



**STUDY:** Howard and Corsi  
**Study year:** 1997  
**Solution Method:** Method 4 for toluene, ethylbenzene, and cyclohexane  
Method 5 for acetone and ethyl acetate

**Assumptions:** None

**Comments:**

\* = unable to be determined with available data

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Cycle Type =	Rinse	H <sub>c</sub> @ 24 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>f</sub>	Mass Closure	
Duration (min) =	10	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	24	Acetone	0.0012	1.1 x 10 <sup>-5</sup>	0.11	36	34	0.013	0.026	7.1	0.069	40	57	1.4	98
Liquid Volume (L) =	47	Ethyl Acetate	0.0048	9.5 x 10 <sup>-6</sup>	0.092	23	22	0.01	0.047	12	0.15	23	32	1.4	106
Gas Flowrate (L/min) =	53	Toluene	0.26	9.1 x 10 <sup>-6</sup>	0.085	5.1	1.4	0.0033	0.091	72	9.4	34	49	1.4	70
Final Headspace Volume (L) =	103	Ethylbenzene	0.31	8.4 x 10 <sup>-6</sup>	0.077	4.1	0.96	0.0024	0.056	76	10	34	49	1.4	75
Agitation Speed =	slow	Cyclohexane	7.1	9.0 x 10 <sup>-6</sup>	0.088	1.4	0.012	0.0019	0.0038	99	24	26	37	1.4	28
Clothes Present =	no														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Cycle Type =	Rinse	H <sub>c</sub> @ 22 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>f</sub>	Mass Closure	
Duration (min) =	10	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	22	Acetone	0.0011	1.1 x 10 <sup>-5</sup>	0.11	44	37	0.013	0.025	15	0.024	30	22	0.74	95
Liquid Volume (L) =	49	Ethyl Acetate	0.0044	9.5 x 10 <sup>-6</sup>	0.092	27	25	0.013	0.057	8.1	0.073	23	17	0.74	103
Gas Flowrate (L/min) =	53	Toluene	0.25	9.1 x 10 <sup>-6</sup>	0.085	7.1	2.5	0.0042	0.19	65	7.1	46	34	0.74	63
Final Headspace Volume (L) =	101	Ethylbenzene	0.28	8.4 x 10 <sup>-6</sup>	0.077	7.1	2.2	0.0033	0.13	69	8.1	47	35	0.74	53
Agitation Speed =	slow	Cyclohexane	6.7	9.0 x 10 <sup>-6</sup>	0.088	3.6	0.028	0.013	0.019	99	23	27	20	0.74	42
Clothes Present =	no														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Cycle Type =	Rinse	H <sub>c</sub> @ 49 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>f</sub>	Mass Closure	
Duration (min) =	10	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	49	Acetone	0.0046	1.1 x 10 <sup>-5</sup>	0.11	48	31	0.024	0.029	36	0.30	62	67	1.1	104
Liquid Volume (L) =	48	Ethyl Acetate	0.014	9.5 x 10 <sup>-6</sup>	0.092	28	15	0.014	0.032	48	0.61	41	45	1.1	114
Gas Flowrate (L/min) =	200	Toluene	0.53	9.1 x 10 <sup>-6</sup>	0.085	8.0	0.38	0.0035	0.0061	95	15	41	44	1.1	112
Final Headspace Volume (L) =	102	Ethylbenzene	1.0	8.4 x 10 <sup>-6</sup>	0.077	7.1	0.24	0.0024	0.0033	97	17	33	36	1.1	97
Agitation Speed =	slow	Cyclohexane	15	9.0 x 10 <sup>-6</sup>	0.088	2.3	0.003	0.0010	1.9 x 10 <sup>-4</sup>	100	46	49	53	1.1	72
Clothes Present =	no														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Cycle Type =	Wash	H <sub>c</sub> @ 23 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>f</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>f</sub>	Mass Closure	
Duration (min) =	9.75	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature ( C) =	23	Acetone	0.0011	1.1 x 10 <sup>-5</sup>	0.11	31	29	0.0093	0.0073	7.0	0.011	13	10	0.74	95
Liquid Volume (L) =	49	Toluene	0.26	9.1 x 10 <sup>-6</sup>	0.085	5.6	3.7	0.0089	0.032	33	1.5	9.6	7.1	0.74	93
Gas Flowrate (L/min) =	53	Ethylbenzene	0.29	8.4 x 10 <sup>-6</sup>	0.077	4.9	3.0	0.0062	0.023	36	2.2	12	9.2	0.74	91
Final Headspace Volume (L) =	101	Cyclohexane	6.7	9.0 x 10 <sup>-6</sup>	0.088	2.0	0.36	0.0061	0.025	82	9.4	11	8.4	0.74	61
Agitation Speed =	slow														
Clothes Present =	no														













**DISHWASHER DATABASE**

**STUDY:** Howard and Corsi  
**Study year:** 1997  
**Solution Method:** Method 4  
**Assumptions:**  $C_{g,in} = 0$ , unless otherwise given  
**Comments:** # = initial liquid-phase concentration based on average of duplicate samples with a relative difference greater than 20%, but no more than 36%.  
 Mass closure is average over four time periods within experiment.  
 \* = unable to be determined with available data

Entry #	Operating Conditions		Chemical Properties				Chemical Concentrations				Mass Transfer Parameters					
1	Cycle Type =	Rinse	$H_c @ 43\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
	Liquid Temperature ( C) =	43	Acetone	0.0034	$1.1 \times 10^{-5}$	0.11	89	45	0	0.25	50 <sup>#</sup>	7.0	*	*	*	95
	Liquid Volume (L) =	7.4	Toluene	0.46	$9.1 \times 10^{-6}$	0.085	16	0.55	0	0.26	97 <sup>#</sup>	33	*	*	*	90
	Headspace Volume (L) =	181	Ethylbenzene	0.80	$8.4 \times 10^{-6}$	0.077	11	0.36	0	0.18	97 <sup>#</sup>	31	*	*	*	84
Ventilation Rate (L/min) =	5.7	Cyclohexane	13	$9.0 \times 10^{-6}$	0.088	5.5	0.016	0	0.075	100 <sup>#</sup>	45	48	*	*	95	
Dishes Present =	No															
2	Cycle Type =	Rinse	$H_c @ 42\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
	Liquid Temperature ( C) =	42	Acetone	0.0032	$1.1 \times 10^{-5}$	0.11	74	49	0	0.22	34	4.2	*	*	*	94
	Liquid Volume (L) =	7.4	Toluene	0.44	$9.1 \times 10^{-6}$	0.085	14	0.63	0	0.27	96	30	*	*	*	91
	Headspace Volume (L) =	181	Ethylbenzene	0.75	$8.4 \times 10^{-6}$	0.077	12	0.41	0	0.18	97	32	*	*	*	86
Ventilation Rate (L/min) =	5.7	Cyclohexane	12	$9.0 \times 10^{-6}$	0.088	7.1	0.018	0	0.070	100	49	49	*	*	82	
Dishes Present =	Yes															
3	Cycle Type =	Rinse	$H_c @ 39\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
	Liquid Temperature ( C) =	39	Acetone	0.0028	$1.1 \times 10^{-5}$	0.11	66	36	0	0.20	43	5.8	*	*	*	93
	Liquid Volume (L) =	7.4	Toluene	0.42	$9.1 \times 10^{-6}$	0.085	8.9	0.27	0	0.17	97	32	*	*	*	91
	Headspace Volume (L) =	181	Ethylbenzene	0.68	$8.4 \times 10^{-6}$	0.077	8.7	0.19	0	0.13	98	35	*	*	*	87
Ventilation Rate (L/min) =	5.7	Cyclohexane	12	$9.0 \times 10^{-6}$	0.088	3.9	0.004	0	0.043	100	58	58	*	*	83	
Dishes Present =	Yes															
4	Cycle Type =	Wash	$H_c @ 43\text{ C}$	$D_l @ 24\text{ C}$	$D_g @ 24\text{ C}$	$C_{l,in}$	$C_{l,out}$	$C_{g,in}$	$C_{g,end}$	$\eta$	$K_L A$	$k_f A$	$k_g A$	$k_g/k_l$	Mass Closure	
	Duration (min) =	10	Chemical	$(m^3_{liq}/m^3_{gas})$	$(cm^2/sec)$	$(cm^2/sec)$	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
	Liquid Temperature ( C) =	43	Acetone	0.0034	$1.1 \times 10^{-5}$	0.11	53	33	0	0.30	37	5.1	*	*	*	92
	Liquid Volume (L) =	7.4	Toluene	0.46	$9.1 \times 10^{-6}$	0.085	10	0.46	0	0.33	96	30	*	*	*	99
	Headspace Volume (L) =	181	Ethylbenzene	0.81	$8.4 \times 10^{-6}$	0.077	10	0.81	0	0.27	97	33	*	*	*	93
Ventilation Rate (L/min) =	5.7	Cyclohexane	13	$9.0 \times 10^{-6}$	0.088	5.1	13	0	0.11	100 <sup>#</sup>	51	51	*	*	87	
Dishes Present =	No															



Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Cycle Type =	Wash	H <sub>c</sub> @ 55 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	6.67	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature (C) =	55	Acetone	0.0061	1.1 x 10 <sup>-5</sup>	0.11	44	28	0.052	0.48	37	4.9	*	*	*	100
Liquid Volume (L) =	7.4	Toluene	0.62	9.1 x 10 <sup>-6</sup>	0.085	9.9	0.34	0.010	0.29	97	31	*	*	*	98
Headspace Volume (L) =	181	Ethylbenzene	1.4	8.4 x 10 <sup>-6</sup>	0.077	10	0.25	0.008	0.21	97	34	*	*	*	90
Ventilation Rate (L/min) =	5.7	Cyclohexane	18	9.0 x 10 <sup>-6</sup>	0.088	5.1	0.014	0.004	0.080	100	47	46	*	*	85
Dishes Present =	Yes														

Operating Conditions		Chemical Properties			Chemical Concentrations				Mass Transfer Parameters						
Cycle Type =	Wash	H <sub>c</sub> @ 53 C	D <sub>l</sub> @ 24 C	D <sub>g</sub> @ 24 C	C <sub>l,in</sub>	C <sub>l,out</sub>	C <sub>g,in</sub>	C <sub>g,end</sub>	η	K <sub>L</sub> A	k <sub>l</sub> A	k <sub>g</sub> A	k <sub>g</sub> /k <sub>l</sub>	Mass Closure	
Duration (min) =	6.67	Chemical	(m <sup>3</sup> <sub>liq</sub> /m <sup>3</sup> <sub>gas</sub> )	(cm <sup>2</sup> /sec)	(cm <sup>2</sup> /sec)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(%)	(L/min)	(L/min)	(L/min)	(%)	
Liquid Temperature (C) =	53	Acetone	0.0056	1.1 x 10 <sup>-5</sup>	0.11	60	36	0.008	0.48	40 <sup>#</sup>	5.2	*	*	*	97
Liquid Volume (L) =	7.4	Toluene	0.6	9.1 x 10 <sup>-6</sup>	0.085	15	0.48	0.003	0.26	97 <sup>#</sup>	35	*	*	*	88
Headspace Volume (L) =	181	Ethylbenzene	1.3	8.4 x 10 <sup>-6</sup>	0.077	17	0.41	0.0020	0.19	98 <sup>#</sup>	37	*	*	*	82
Ventilation Rate (L/min) =	5.7	Cyclohexane	17	9.0 x 10 <sup>-6</sup>	0.088	8.5	0.026	0	0.092	100 <sup>#</sup>	55	54	*	*	84
Dishes Present =	Yes														